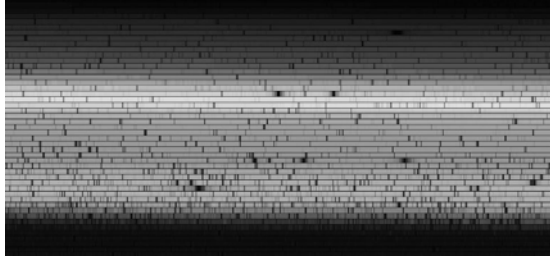


ASTR 1120: Stars & Galaxies



Prof. Juri Toomre TA: Ben Brown

Lecture 4 Wed 19 Jan 05

zeus.colorado.edu/astr1120-toomre

Topics for Today

- Recall: as electrons pop from one level to another, yield 'spectral lines': fingerprints are unique to each atom
- Kirchoff's "laws" + black-body emission spectrum
- Doppler effect
- Begin with telescopes

- *Observatory Night #2 tomorrow*, Thur Jan 20 (at Sommers-Bausch Observ, next to Fiske and Kittredge) by sign-up (7pm, 8pm, 9pm)
- Forecast looks GOOD: Clear and warm!

Reminder Homework Set 1

- *Part A* involves going to book website, after login 'joining our class' (cm228574 as in syllabus), doing the 'Light & Spectroscopy' tutorial in Chap 6 while having your performance e-recorded (can repeat as often as wish). Complete by classtime this Friday. Web server for *astronomyplace* gets a bit clogged.
- *Part B* involves completing the 'Energy Level Diagrams & Spectral Lines' problem sheet passed out in class last Fri. Due this Friday in class, no lates. Show how you got answers for Q 6-8 by staple-attaching worksheet.

Reading for Next Class

- Finish reading Chap 7, *Telescopes*
- Start reading Chap 15, *The Sun*

- *Check with us if any problems with your clickers*
- *Come see us if you need any help or advice with Homework Set 1, or with making your accounts active or with finding your way on www.astronomyplace.com*

Reading Clicker Q **B**

What really is a light year ?

- **A.** Year that has less calories
- **B.** Distance travelled by light in a year
- **C.** Time before some politicians tell the truth (as in "It will be many light years before the truth is known..")
- **D.** Travel time for light to get to us from the Sun

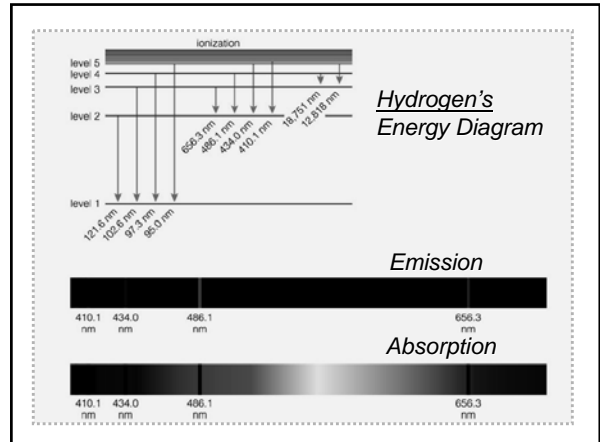
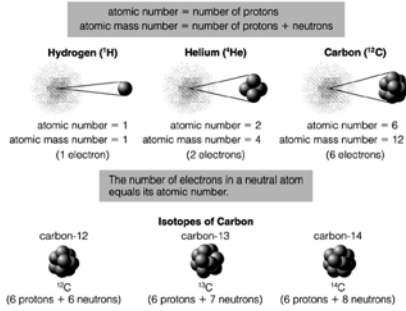
Light year ?

- **B.** DISTANCE travelled if moving at the speed of light (300,000 km/sec)

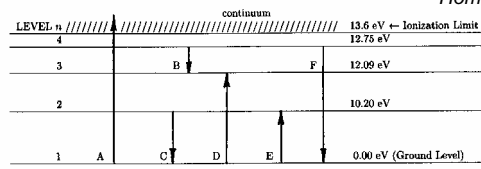
1 light-year = 9.46 trillion km = $9.46 \cdot 10^{12}$ km

Light travel time from surface of Sun to us is about 8 minutes

Nucleus and its electron cloud



Below is an energy level diagram for hydrogen, with one electron. The capital letters represent individual atoms undergoing the indicated electronics transitions (jumping up or down between orbits). **Homework:1**



$$E = hf \quad c = \lambda f$$

Where: E is photon energy (eV) f is photon frequency ($\text{sec}^{-1} = \text{Hz}$) h (Planck's constant) = $4.14 \times 10^{-15} \text{ eV sec}$ c (speed of light) = $3 \times 10^{10} \text{ cm/sec}$ $1 \text{ \AA} = 1 \times 10^{-8} \text{ cm}$ λ (lambda) is photon wavelength (cm or \AA)


Using the Hydrogen atom energy diagram given above, the equations and the constants as listed, calculate the wavelength in Angstroms (\AA) of the following transitions:


6. Transition E? _____ Transition B? _____


Advice: Read Appendix C.3 "Working with Units"

NATURE OF SPECTRA

SPECTRA (KIRCHOFF'S LAWS)

A.  **RADIATING SOLID, LIQUID, OR HIGH-PRESSURE GAS (ORIGINALLY THICK)**
⇒ **CONTINUOUS SPECTRUM**

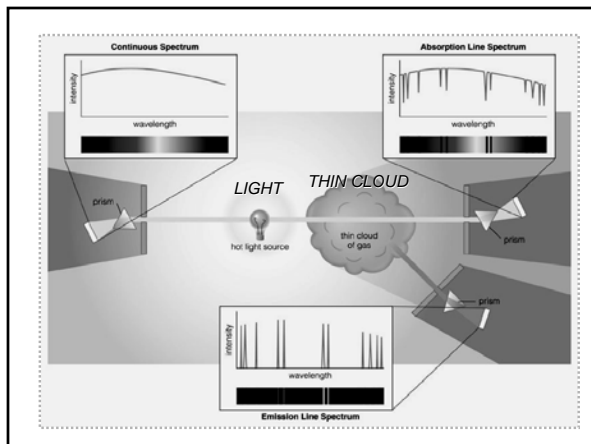
B.  **RADIATING RARIFIED GAS (LOW DENSITY)**
⇒ **BRIGHT-LINE SPECTRUM (EMITTON)**

C.  **COOL** ⇒ **LIGHT OF CONTINUOUS SPECTRUM VIEWED THROUGH A COOLER GAS**
⇒ **DARK-LINE (ABSORPTION) SPECTRUM**

"Kirchoff's Laws"

Emission

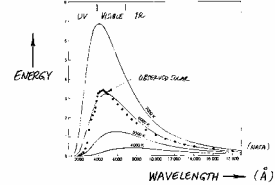
Absorption



CONTINUOUS SPECTRUM OF EMISSION BY "BLACK BODY"

PLANCK'S RADIATION LAW: BEHAVIOR OF THERMAL RADIATION AS TEMPERATURE IS VARIED

BLACK-BODY SPECTRUM



Planck

Peak emission and total energy VARY with temperature

TOTAL ENERGY EMITTED $\sim (\text{TEMPERATURE})^4$

$$E = \sigma T^4 \quad \text{STEFAN-BOLTZMANN LAW}$$

Stefan-Boltzmann

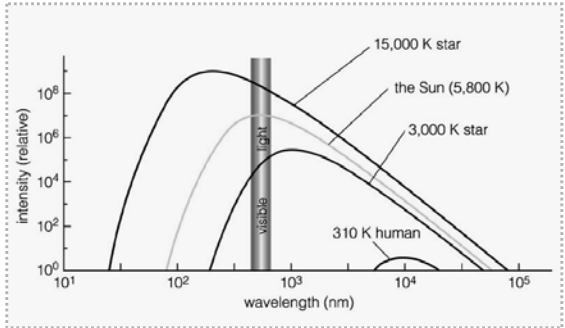
WAVELENGTH OF PEAK EMISSION $\sim 1 / \text{TEMPERATURE}$

$$\lambda_{\text{MAX}} \sim \frac{1}{T} \quad \text{WIEN'S LAW}$$

Wien

{ IF INCREASE T FROM 1000 K TO 6000 K, E INCREASES BY FACTOR $6 \times 6 \times 6 \times 6 = 1296!$ }

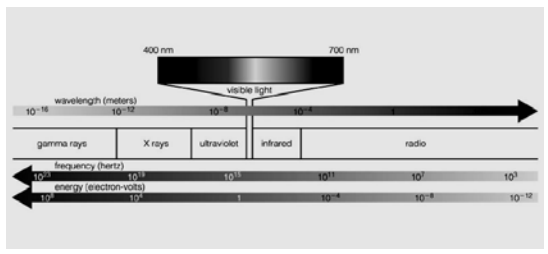
Spectra of Continuous Emission from Us and Stars



Clicker Q – EM Waves

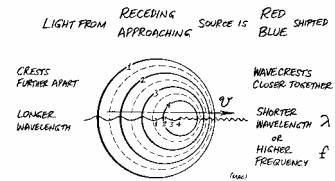
- From *shortest to longest wavelength*, what is the correct sequence of EM radiation?
- A. gamma-rays, x-rays, UV, visible, IR, radio
- B. gamma-rays, x-rays, visible, UV, IR, radio
- C. IR, visible, UV, x-rays, gamma-rays, radio
- D. radio, IR, visible, UV, x-rays, gamma-rays

A. gamma-rays, x-rays, UV, visible, IR, radio



DOPPLER EFFECT

DOPPLER EFFECT



$$\text{CHANGE IN WAVELENGTH} = \frac{\Delta \lambda}{\lambda} = \frac{v}{c} = \frac{\text{VELOCITY OF SOURCE}}{\text{SPEED OF LIGHT}}$$

Applied to positions of spectral lines

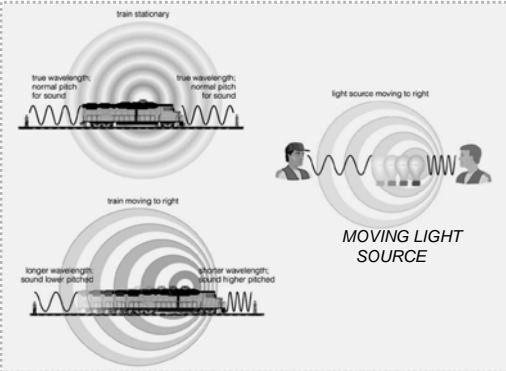
CAN USE TO CALCULATE LINE-OF-SIGHT VELOCITY OF SOURCE: "DOPPLER VELOCITY" v

$$v = \frac{\Delta \lambda}{\lambda} c$$

IF ABSORPTION LINE AT SAY 5000 Å REDSHIFTED BY 0.5 Å

$$v = \frac{(+0.5 \text{ \AA})}{5000 \text{ \AA}} (300,000 \text{ km/sec}) = +30 \text{ km/sec}$$

Doppler Effect: Trains and Light



Now On to Telescopes



VLA – Scicorro, NM